## Synthesis of Partially Saturated N-Substituted 4*H*-3,1-Benzothiazine-2(1*H*)-thiones

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Summary. Acid-catalyzed reaction of 2-arylidenecyclohexanones 1 with N-substituted dithiocarbamic acids 2 gave open-chain addition products 3 and 4. Dehydration of 3 and 4 afforded only one of the three possible isomeric N-substituted 4H-3,1-benzothiazine-2(1H)-thiones 5 and 6.

Keywords. N-Substituted 4H-3,1-benzothiazine-2(1H)-thiones; Synthesis; <sup>1</sup>H- and <sup>13</sup>C-spectroscopy.

## Synthese von partiell gesättigten N-substituierten 4H-3,1-Benzothiazin-2-(1H)-thionen

**Zusammenfassung.** Die säurekatalysierte Reaktion von 2-Arylidencyclohexanonen 1 mit N-substituierten Dithiocarbaminsäure 2 ergab die offenkettigen Additionsprodukte 3 und 4. Die Dehydratation von 3 und 4 führte ausschließlich zu einem der drei möglichen N-substituierten 4H-3,1-Benzothiazin-2(1H)-thion-Isomeren 5 und 6.

## Introduction

The chemistry and pharmacology of 3,1-benzothiazines have scarcely been investigated. The synthetic works were focused on preparation and simple transformation of the compounds. Transformation of 4-aryl-3,1-benzothiazine-2-thiones into 4,1benzothiazepine derivatives is one of the most important results of the latter studies [1]. As a result of the pharmacological investigations, a few 4-aryl-2-(thi)oxo-, and 4-aryl-2-amino-3,1-benzothiazine derivatives with CNS [2–4], analgesic [5, 6], and antimicrobial [7, 8] effects were reported.

Earlier, we have shown that reaction of dithiocarbamic acid with 2-arylidenecyclohexanones is a versatile route for synthesis of 4-aryl-4H-3,1-benzothiazine-2(1H)-thiones [9]. As a continuation of our earlier work, we report here the results obtained by treating 2-arylidenecyclohexanones 1 a-f with N-substituted dithiocarbamic acids 2 a and 2 b. In these reactions formation of partially saturated N-substituted 4H-3,1-benzothiazine-2(1H)-thiones was expected, which were considered to be the first representatives of N-substituted 3,1-benzothiazines.

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## **Results and Discussion**

The reaction of 2-arylidenecyclohexanones 1 **a**–f with dithiocarbamic acids 2 **a** and 2 **b** was carried out in acidic aqueous acetone solutions at  $-5^{\circ}$ C to yield the openchain addition products 3 **a**–f and 4 **a**–d (Scheme 1). The appearance of the v<sub>(NH)</sub> and v<sub>(C=O)</sub> signals in their IR spectra proved unambiguous evidence of the progress of the addition reactions as well as the open-chain structure of the products. The <sup>1</sup>H-NMR (60 MHz, *DMSO-d*<sub>6</sub>) spectra were rather complex suggesting equilibrium between the open-chain and cyclic N,O-hemiketal structures [9] of the compounds.



In order to obtain the expected 3,1-benzothiazines we tried to apply the dehydration methods (Method A: p-TSA/Benzene; Method B: TFA/Benzene; Method C:  $(C_2H_5)_2O - BF_3/CHCl_3$ ) used in our earlier work, which proved to be slightly selective in dehydration of the configurationally different cyclic dithiocarbamic acid adducts of 2-arylidenecyclohexanones [9]. These methods, however, failed, even if longer reaction times were used. Thus, compounds **3 a**-**f** and **4 a**-**d** were dehydrated in acetic anhydride using sulphuric acid as catalyst. <sup>1</sup>H-NMR (60 MHz, CDCl<sub>3</sub>) analysis of the crude reaction products showed that dehydration of **3 a**-**f** and

Compound	$CH_2 (5, 6)$ $2 \times m (2 \times 2H)$	CH <sub>2</sub> (7) m (2 H)	H-4a m (1 H)	H-4 d (1 H) <sup>b</sup>	$NCH_2^c$ 2 × d/m (	2×1H)	$H-8 \sim t (1 H)^d$	ArH (Pos. 4 + side chain) m's/d's (9/10 H) <sup>e</sup>
5a	~ 1.25, ~ 1.45	2.05	2.98 <sup>f</sup>	4.64	5.62,	5.86	5.77	7.2-7.5
5b	$\sim 1.4, \sim 1.5$	2.10	2.82	4.35	5.46,	5.99	5.66	$6.88^{\text{g}}, 7.2-7.4^{\text{h}}$
5c	$\sim 1.45, \sim 1.55$	2.10	2.85	4.35	5.46,	6.00	5.66	7.1-7.4
5 d	~ 1.3, ~ 1.5	2.08	3.00	4.71	5.60,	5.86	5.79	$\sim$ 7.25 <sup>h</sup> , $\sim$ 7.35 <sup>h</sup> , 7.46 <sup>g</sup> , 7.52 <sup>g</sup>
5e	$\sim 1.3, \sim 1.5$	2.05	3.02	4.62	5.58,	5.88	5.76	$6.92^{i}$ , $7.02^{j}$ , $7.2-7.4^{h}$
5f	$\sim 1.35, \sim 1.5$	2.10	3.08	4.35	5.46,	5.95	5.68	7.2–7.4 <sup>h</sup> , 7.47 <sup>g</sup>
6a	1.15- 1.6	2.15	$\sim 2.6$	4.52	$\sim 4.55, -$	~ 4.8	6.00	7.2–7.4
6b	1.2 - 1.6	2.15	2.55	4.45	$\sim 4.55, c$	~ 4.8	5.99	$6.93^{\mathrm{g}}, \sim 7.3^{\mathrm{h}}$
6c	1.1- 1.6	2.15	2.55	4.48	~ 4.4, ~	~ 4.8	6.04	$\sim$ 7.2 <sup>h</sup> , $\sim$ 7.3 <sup>h</sup>
6d	1.1 - 1.6	2.15	$\sim 2.6$	4.57	$\sim 4.2, $	~ 4.8	6.02	$\sim 7.2 - 7.5$

<sup>a</sup> Solvent  $DMSO-d_6$  for 5a, d, f and 6a-d,  $CDCl_3$  for 5b, c, e

<sup>b</sup> J (H-4, H-4a): 11.7  $\pm$  0.2

<sup>c</sup> AB-type spectrum (2 × d) for 5a-f, J (A, B): 16.0  $\pm$  0.2, AB-part (2 × m) of an ABXY-system for 6a-d

<sup>d</sup> J:  $4.0 \pm 0.1$ 

<sup>c</sup> Total intensity 10 H (5 a, 6 a), 9 H (5 b-f, 6 b-d)

 $^{\rm f} \sim {\rm d} \ (J: \ 11.3)$ 

<sup>g</sup> A or B part (intensity: 2H) of the AA'BB'-type multiplet of the *para*-disubstituted aromatic ring (Pos. 4), J (A, B): 8.8 (5b, 6b),  $\sim 9$  (5d), 8.4 (5f)

<sup>h</sup> Overlapping m's, intensity: 7 H (5 b, f, 6 b), 3 + 2 H (5 d), 6 H (5 e), 5 + 4 H (6 c)

<sup>i</sup> H-4 (4-aryl group), dd (1 H)

<sup>j</sup> H-2, 6 (4-aryl group), overlapping d's (2 H)

C4°	C4	C – C°	C1	NCH, <sup>b</sup>	C-8a	C-8	C-6	C-5. 7	C-4a	$C-4^{b}$	C = S(2)	Compound
t (C-3°)	and 161.4	-2°), 122.2 (C-6°	at 115.9 (C-	urther ones a	and three f	given above	he interval	r 5e, three in t	ur <sup>c</sup> lines fo	o plus for	There are two	(5f, C-3, 5).
l given above $.7$ and $132.2^{\circ}$	the interval b) and 126	one line out of $5$ ) and $126.7$ ( $5$	earance of 4.5° (C-3,	a) or of appe or 5b, f at 11	nd 130.0 (6) separated fo	130.5 (5a) a nd two are	vo lines at e interval a	of overlap of tware in the abov	It because of Two lines	and <b>6 a</b> , 6 C-3, 5).	nes lor <b>5a, c</b> 116.0° ( <b>6b</b> ,	<b>6 c, d</b> , three II at 126.8 (5 c),
m for <b>5 d</b> and given above	id 132.0 ppi the interval	etween 128.0 an one line out of	four lines b earance of	4r-2, 3, 5, 6: 1 a) or of appe	.0 (6 c, d); C ind 130.0 (6;	2 (6 a, b), 34 130.5 (5 a) a	chain): 34. vo lines at	); <i>Ph</i> CH <sub>2</sub> (side of two setups of the setup of the setu	), 22.3 (6 c d because (	, 54.9 (6b and 6a,	:), 56.9° (5 e), nes for 5 a, c	(5b), 21.1 (5c 6c, d, three lii
CH <sub>3</sub> (X): 55.3	er signals: (	53 MHz <sup>a</sup> . Furthe	n at 20 or (	DCl <sub>3</sub> solution	-d <sub>6</sub> and/or C	d in DMSO	⊢f and 6a⊣	compounds 5 a	(g/ppm) of	al shifts (	NMR chemic	<b>Table 2.</b> <sup>13</sup> C-1

Compound	C = S(2)	C-4 <sup>b</sup>	C-4 a	C-5, 7	C-6	C-8	C-8a	NCH <sub>2</sub> <sup>b</sup>	$C_{Ar}$ -1	$C_{Ar} - C^{\circ}$	$C_{Ar}$ -4	$C_{Ar}$ -4°
5 a	192.4	55.2	40.2	26.0, 27.4	19.6	118.8	140.1	56.9	138.0 <sup>b</sup>	138.1 <sup>b</sup>	128.5	129.9
5b	192.5	54.3	39.9	24.9, 26.2	18.6	117.0	139.4	56.4	136.5	128.2	127.1	159.7
5c	192.7	54.7	39.9	25.0, 26.4	18.7	116.9	139.6	56.4	133.5	136.6	127.2	138.4
5d	192.1	54.4	40.0	26.0, 27.4	19.5	119.0	139.9	56.9	137.3 <sup>b</sup>	137.9 <sup>b</sup>	128.6	134.0
5e	192.4	55.1	40.1	26.0, 27.5	19.6	118.7	140.2	56.9 <sup>d</sup>	138.0	139.7	128.5	115.4
5f	191.8	54.3	39.7	24.8, 26.3	18.6	117.3	139.0	56.4	135.7	136.3	127.2	122.4
6a	191.0	55.2	39.7	26.0, 26.9	19.2	119.0	$140.3^{b}$	55.4	138.5	$139.9^{\mathrm{b}}$	128.0	129.8
6 b	191.3	55.2	39.9	26.0, 26.9	19.2	118.8	140.5	56.8	139.9	130.2	128.0	160.8
6c	191.0	55.2	39.5	25.9, 26.8	19.0	119.0	139.9 <sup>b</sup>	55.0	140.3	135.5	128.0	139.2 <sup>b</sup>
6d	190.6	55.1	39.6	25.8, 26.9	19.0	119.1	140.0	55.4	139.0	137.6	127.8	134.4

<sup>a</sup> Solvent:  $DMSO-d_6$  (5 a, d, e, and 6 a–d) or CDCl<sub>3</sub> (5 b, c, f, and 6 c, d), measuring frequency: 20.14 MHz (5 b, c, f) or 62.89 MHz (5 a, d, e, and 6 a–d) <sup>b</sup> Interchangeable assignments

° 4-aryl group <sup>d</sup> Two overlapping lines

4a-d furnished only one of the three possible 3,1-benzothiazine isomers 5a-f and 6a-d (Scheme 1) [10].

The <sup>1</sup>H- and <sup>13</sup>C-NMR data of compounds 5a-f and 6a-d are given in Tables 1 and 2 and provide convincing evidence of the structures of the new products. The values (ca. 11.7 Hz) of J (H-4, H-4a) coupling constants prove [11] the 1,3diaxial arrangements of the hydrogens involved and consequently the *cis*-equatorial position of the 4-aryl group relative to the H-4a and the heteroring, respectively. Both 7-methylen hydrogens have couplings of the same magnitude to H-8 (the olefinic signal is a pseudo-triplet split by ca 4 Hz). All evidence suggests a preferred conformation for the flexible compounds, with the alicycle in half-chair form (where C-6 and C-7 have "up and down" position to the plane of C-4a, 7, 8, 8a atoms) and the hetero ring in a twisted-boat form (Fig. 1).

## **Experimental Part**

Melting points were determined on a Boetius apparatus and are uncorrected. IR spectra were taken in KBr pellets with a Specord 75 IR spectrophotometer. <sup>1</sup>H-NMR spectra were recorded with a Perkin-Elmer R-12 equipment (60 MHz) or a Bruker WM-250 FT-spectrometer, using *TMS* as internal standard, at 35°C or 25°C, respectively. <sup>13</sup>C-NMR measurements were carried out at 25°C on Bruker WM-250 or WP-80 SY FT-spectrometers, at 63 or 20 MHz, respectively, using *TMS* as internal standard. Elemental analyses were performed in-house and at the Central Research Laboratory, University Medical School, Pécs.

2-Arylidenecyclohexanones [12] and N-substituted dithiocarbamic acids [13] used as starting materials were synthesized by literature methods. The (*E*)-configuration of the unsaturated ketones was based on <sup>1</sup>H-NMR investigations [14].

The isomeric composition of the reaction products was examined by <sup>1</sup>H-NMR spectroscopy (60 MHz), based on investigation of the well separated H-4 signals. TLC was performed on Kieselgel GF 254 plates (Merck) using benzene as eluant.

# General Procedure for the Addition of Dithiocarbamic Acids 2a and 2b to 2-Arylidenecyclohexanones 1a-f

To a solution of 0.075 mol of ammonium salt of N-benzyl- (2 a), or N-(2-phenylethyl)-dithiocarbamic acid (2 b) dissolved in 150 ml of 50% methanol (cooled to  $-5^{\circ}$ C), 35 ml of 6.5N hydrochloric acid (cooled to  $-5^{\circ}$ C) was added dropwise while stirring. Cooling and stirring were continued, and 0.035 mol of unsaturated ketone 1 a-f in 200 ml acetone (cooled to  $-5^{\circ}$ C) was added to the reaction mixture. After stirring at this temperature for 4 h, the precipitate formed was filtered off, washed free of acid with water, dried, and crystallized from benzene/petroleum ether to give colourless crystals.

Fig. 1. Conformation of compounds 5 and 6

## 2-(a-(N-Benzyl-thiocarbamoylthio)-benzyl)-cyclohexan-1-one (3 a)

Yield: 83%, m.p. 108–111°C. IR (KBr):  $v = 3320 \text{ cm}^{-1}$  (NH), 2925, 2945 cm<sup>-1</sup> (CH<sub>2</sub>), 1690 cm<sup>-1</sup> (C=O). C<sub>21</sub>H<sub>23</sub>NOS<sub>2</sub> (369.54). Calcd. C 68.26, H 6.27, S 17.35; found C 68.34, H 6.21, S 17.10.

## 2-(a-(N-Benzyl-thiocarbamoylthio)-4-methoxybenzyl)-cyclohexan-1-one (3b)

Yield: 81%, m.p. 106–109°C. IR (KBr):  $v = 3.275 \text{ cm}^{-1}$  (NH), 2.930 cm<sup>-1</sup> (CH<sub>2</sub>), 1.695 cm<sup>-1</sup> (C=O). C<sub>22</sub>H<sub>25</sub>NO<sub>2</sub>S<sub>2</sub> (399.57). Calcd. C 66.13, H 6.31, S 16.05; found C 66.24, H 6.18, S 15.80.

#### 2-(a-(N-Benzyl-thiocarbamoylthio)-4-methylbenzyl)-cyclohexan-1-one (3 c)

Yield: 78%, m.p. 100–104°C. IR (KBr):  $v = 3.335 \text{ cm}^{-1}$  (NH), 2930, 2945 cm<sup>-1</sup> (CH<sub>2</sub>), 1700 cm<sup>-1</sup> (C=O). C<sub>22</sub>H<sub>25</sub>NOS<sub>2</sub> (383.57). Calcd. C 68.89, H 6.57, S 16.72; found C 68.64, H 6.78, S 16.67.

## 2-(a-(N-Benzyl-thiocarbamoylthio)-4-chlorobenzyl)-cyclohexan-1-one (3d)

Yield: 73%, m.p. 110–113°C. IR (KBr):  $v = 3\,300\,\text{cm}^{-1}(\text{NH})$ , 2940 cm<sup>-1</sup> (CH<sub>2</sub>), 1700 cm<sup>-1</sup> (C=O). C<sub>21</sub>H<sub>22</sub>ClNOS<sub>2</sub> (403.98). Calcd. C 62.44, H 5.49, S 15.87; found C 62.23, H 5.37, S 15.98.

## 2-(a-(N-Benzyl-thiocarbamoylthio)-3-methoxybenzyl)-cyclohexan-1-one (3e)

Yield: 85%, m.p. 103–105°C. IR (KBr):  $v = 3.225 \text{ cm}^{-1}$ (NH), 2.940 cm<sup>-1</sup> (CH<sub>2</sub>), 1.695 cm<sup>-1</sup> (C=O). C<sub>22</sub>H<sub>25</sub>NO<sub>2</sub>S<sub>2</sub> (399.57). Calcd. C 66.13, H 6.31, S 16.05; found C 66.04, H 6.48, S 15.94.

#### 2-(a-(N-Benzyl-thiocarbamoylthio)-4-bromobenzyl)-cyclohexan-1-one (3 f)

Yield: 86%, m.p. 116–118°C. IR (KBr):  $v = 3.305 \text{ cm}^{-1}$ (NH),  $2.935 \text{ cm}^{-1}$  (CH<sub>2</sub>),  $1.700 \text{ cm}^{-1}$  (C=O). C<sub>21</sub>H<sub>22</sub>BrNOS<sub>2</sub> (448.43). Calcd. C 56.25, H 4.94, S 14.30; found C 56.07, H 4.71, S 14.51.

2-(a-(N-(2-Phenylethyl)-thiocarbamoylthio)-benzyl)-cyclohexan-I-one (4 a)

Yield: 80%, m.p. 119–122°C. IR (KBr):  $v = 3.355 \text{ cm}^{-1}$ (NH), 2945 cm<sup>-1</sup> (CH<sub>2</sub>), 1690 cm<sup>-1</sup> (C=O). C<sub>22</sub>H<sub>25</sub>NOS<sub>2</sub> (383.57). Calcd. C 68.89, H 6.57, S 16.72; found C 68.72, H 6.64, S 16.90.

#### 2-(a-(N-(2-Phenylethyl)-thiocarbamoylthio)-4-methoxybenzyl)-cyclohexan-1-one (4b)

Yield: 87%, m.p. 114–116°C. IR (KBr):  $v = 3.355 \text{ cm}^{-1}(\text{NH})$ , 2940 cm<sup>-1</sup> (CH<sub>2</sub>), 1700 cm<sup>-1</sup> (C=O). C<sub>23</sub>H<sub>27</sub>NO<sub>2</sub>S<sub>2</sub> (413.59). Calcd. C 66.79, H 6.58, S 15.50; found C 66.57, H 6.43, S 15.71.

2-(a-(N-(2-Phenylethyl)-thiocarbamoylthio)-4-methylbenzyl)-cyclohexan-1-one (4 c)

Yield: 85%, m.p. 126–128°C. IR (KBr):  $v = 3.315 \text{ cm}^{-1}$ (NH), 2940 cm<sup>-1</sup> (CH<sub>2</sub>), 1700 cm<sup>-1</sup> (C=O). C<sub>23</sub>H<sub>27</sub>NOS<sub>2</sub> (397.59). Calcd. C 69.48, H 6.84, S 16.13; found C 69.39, H 6.71, S 16.37.

 $2 - (a - (N - (2 - Phenylethyl) - thiocarbamoylthio) - 4 - chlorobenzyl) - cyclohexan - 1 - one (\mathbf{4d})$ 

Yield: 72%, m.p. 126–129°C. IR (KBr):  $v = 3240 \text{ cm}^{-1}(\text{NH})$ , 2930 cm<sup>-1</sup> (CH<sub>2</sub>), 1700 cm<sup>-1</sup> (C=O). C<sub>22</sub>H<sub>24</sub>CINOS<sub>2</sub> (418.01). Calcd. C 63.21, H 5.79, S 15.34; found C 63.07, H 5.63, S 15.47.

#### 4H-3,1-Benzothiazine-2(1H)-thiones

#### General Procedure for Dehydration of Compounds 3 and 4

To the suspension of compounds 3 and 4 (0.03 mol) in 80 ml acetic anhydride 0.5 ml conc. sulphuric acid was added dropwise with stirring. Stirring was continued for  $\frac{1}{2}$  h. Then the mixture was cooled, the precipitate formed was filtered off, washed free of acid with water and dried. The product obtained was subjected to column chromatography (Merck, Kieselgel 60, 0.0063–0.2 mm; benzene) and crystallized from benzene/petroleum ether to give colourless crystals. For the <sup>1</sup>H-NMR spectra of compounds 5 a–f see Table 1, for the corresponding <sup>13</sup>C-NMR see Table 2.

#### trans-N-Benzyl-4-phenyl-4a,5,6,7-tetrahydro-4H-3,1-benzothiazine-2(1H)-thione (5a)

Yield: 93%, m.p. 177–179°C. IR (KBr): v = 2910, 2920, 2945 cm<sup>-1</sup> (CH<sub>2</sub>), 1655 cm<sup>-1</sup> (C=C), 1605, 1495 cm<sup>-1</sup> (C=C<sub>Ar</sub>). C<sub>21</sub>H<sub>21</sub>NS<sub>2</sub> (351.52). Calcd. C 71.75, H 6.02, S 18.24; found C 71.68, H 5.93, S 18.14.

#### trans-N-Benzyl-4-(4-metoxyphenyl)-4a,5,6,7-tetrahydro-4H-3,1-benzothiazine-2(1H)-thione (5b)

Yield: 89%, m.p. 161–163°C. IR (KBr): v = 2910,  $2925 \text{ cm}^{-1}$  (CH<sub>2</sub>),  $1655 \text{ cm}^{-1}$  (C=C), 1605,  $1510 \text{ cm}^{-1}$  (C=C<sub>Ar</sub>). C<sub>22</sub>H<sub>23</sub>NOS<sub>2</sub> (381.55). Calcd. C 69.25, H 6.08, S 16.81; found C 70.05, H 6.19, S 16.68.

#### trans-N-Benzyl-4-(4-methylphenyl)-4a,5,6,7-tetrahydro-4H-3,1-benzothiazine-2(1H)-thione (5c)

Yield: 79%, m.p. 188–190°C. IR (KBr): v = 2910,  $2925 \text{ cm}^{-1}$  (CH<sub>2</sub>),  $1655 \text{ cm}^{-1}$  (C=C), 1605,  $1495 \text{ cm}^{-1}$  (C=C<sub>Ar</sub>). C<sub>22</sub>H<sub>23</sub>NS<sub>2</sub> (365.55). Calcd. C 72.29, H 6.34, S 17.54; found C 72.57, H 6.21, S 17.44.

#### trans-N-Benzyl-4-(4-chlorophenyl)-4a,5,6,7-tetrahydro-4H-3,1-benzothiazine-2(1H)-thione (5 d)

Yield: 84%, m.p.: 183–185°C. IR (KBr):  $v = 2.940 \text{ cm}^{-1}$  (CH<sub>2</sub>), 1.655 cm<sup>-1</sup> (C=C), 1.605, 1.490 cm<sup>-1</sup> (C=C<sub>Ar</sub>). C<sub>21</sub>H<sub>20</sub>ClNS<sub>2</sub> (385.97). Calcd. C 65.35, H 5.22, S 16.61; found C 65.38, H 5.07, S 16.54.

trans-N-Benzyl-4-(3-methoxyphenyl)-4a,5,6,7-tetrahydro-4H-3,1-benzothiazine-2(1H)-thione (5e)

Yield: 89%, m.p. 164–166°C. IR (KBr):  $v = 2910, 2930, 2945 \text{ cm}^{-1}$  (CH<sub>2</sub>), 1655 cm<sup>-1</sup> (C=C), 1605, 1585 cm<sup>-1</sup> (C=C<sub>Ar</sub>). C<sub>22</sub>H<sub>23</sub>NOS<sub>2</sub> (381.55). Calcd. C 69.25, H 6.08, S 16.81; found C 69.41, H 5.83, S 16.64.

#### trans-N-Benzyl-4-(4-bromophenyl)-4a,5,6,7-tetrahydro-4H-3,1-benzothiazine-2(1H)-thione (5f)

Yield: 87%, m.p. 164–166°C. IR (KBr): v = 2915, 2930, 2945 cm<sup>-1</sup> (CH<sub>2</sub>), 1655 cm<sup>-1</sup> (C=C), 1590, 1485 cm<sup>-1</sup> (C=C<sub>Ar</sub>). C<sub>21</sub>H<sub>20</sub>BrNS<sub>2</sub> (430.42). Calcd. C 58.60, H 4.68, S 14.90; found C 58.42, H 4.79, S 15.01.

#### trans-N-(2-Phenylethyl)4-phenyl-4a,5,6,7-tetrahydro-4H-3,1-benzothiazine-2(1H)-thione (6a)

Yield: 91%, m.p. 175–177°C. IR (KBr):  $v = 2.940 \text{ cm}^{-1}$  (CH<sub>2</sub>), 1655 cm<sup>-1</sup> (C=C), 1600, 1495 cm<sup>-1</sup> (C=C<sub>4r</sub>). C<sub>22</sub>H<sub>23</sub>NS<sub>2</sub> (365.55). Calcd. C 72.29, H 6.34, S 17.54; found C 71.97, H 6.41, S 17.64.

trans-N-(2-Phenylethyl)4-(4-methoxyphenyl)-4a,5,6,7-tetrahydro-4H-3,1-benzothiazine-2(1H)-thione (6b)

Yield: 89%, m.p. 171–173°C. IR (KBr):  $v = 2.945 \text{ cm}^{-1}$  (CH<sub>2</sub>), 1640 cm<sup>-1</sup> (C=C), 1610, 1510 cm<sup>-1</sup> (C=C<sub>Ar</sub>). C<sub>23</sub>H<sub>25</sub>NOS<sub>2</sub> (395.58). Calcd. C 69.83, H 6.37, S 16.21; found C 70.04, H 6.19, S 16.37.

trans-N-(2-Phenylethyl)4-(4-methylphenyl)-4a,5,6,7-tetrahydro-4H-3,1-benzothiazine-2(1H)-thione (6 c)

Yield: 81%, m.p. 197–199°C. IR (KBr):  $v = 2.940 \text{ cm}^{-1}$  (CH<sub>2</sub>), 1 640 cm<sup>-1</sup> (C=C), 1 605, 1 515 cm<sup>-1</sup> (C=C<sub>Ar</sub>). C<sub>23</sub>H<sub>25</sub>NS<sub>2</sub> (379.58). Calcd. C 72.78, H 6.64, S 16.89; found C 72.64, H 6.81, S 17.03.

trans-N-(2-Phenylethyl)4-(4-chlorophenyl)-4a,5,6,7-tetrahydro-4H-3,1-benzothiazine-2(1H)-thione (6 d)

Yield: 87%, m.p. 197–199°C. IR (KBr): v = 2910,  $2930 \text{ cm}^{-1}$  (CH<sub>2</sub>),  $1650 \text{ cm}^{-1}$  (C=C), 1600,  $1490 \text{ cm}^{-1}$  (C=C<sub>Ar</sub>). C<sub>22</sub>H<sub>22</sub>ClNS<sub>2</sub> (400.00). Calcd. C 66.06, H 5.54, S 16.03; found C 65.83, H 5.78, S 16.14.

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